

Syntheses of Substituted Succinimides by Radical β -Fragmentation of Bicyclic Carbinol Amides. A New Expedient Synthesis of 2-[2-(Methoxycarbonyl)ethyl]-3-[(methoxycarbonyl)methyl]-3-methylsuccinimide, the Ring-B Imide of Vitamin B₁₂

Rosendo Hernández and Ernesto Suárez*

Instituto de Productos Naturales y Agrobiología del CSIC, Carretera de la Esperanza, 2, 38206-La Laguna, Tenerife, Spain

Daniel Melián

Departamento de Química Orgánica, Universidad de La Laguna, Tenerife, Spain

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A mild and efficient synthesis of substituted succinimides by radical β -fragmentation of carbinol amides is described. The carbinol amides studied were generated by treatment of the corresponding α,β -unsaturated ketones with hydrogen cyanide and subsequent hydrolysis of the cyanide intermediates. Photolysis with visible light in the presence of (diacetoxyiodo)benzene (DIB) and iodine of azabicyclic carbinol amides of the types 1-hydroxy-2-azabicyclo[3.3.0]octan-3-one (e.g., 4, 6, and 7) and 1-hydroxy-2-azabicyclo[3.2.1]octan-3-one (e.g., 16, 17, and 21) led to the substituted succinimides in good yields. A novel and operationally simple synthesis of the ring-B imide 28 is described, employing this β -fragmentation reaction on the carbinol amide 7 as the key step. The transformation of the resulting succinimide 11 into the target 28 was accomplished in one step by oxidation of the primary iodide and the phenyl group with RuO₄.

Five-membered cyclic imides and their N-derivatives are of considerable interest because of their presence in structures of interesting naturally occurring compounds such as the antibiotics isohematinic acid¹ and showdomycin² and synthetic compounds such as azaprostaglandins³ and because of their significance as building blocks for the synthesis of isobacteriochlorins.⁴ They are also useful intermediates, via N-acyliminium ions, α -acylamino radicals, or intramolecular Wittig reactions, for the synthesis of pyrrolidizine alkaloids.⁵

The methods usually used to synthesize cyclic imides involve the formation of a carbon-nitrogen bond from a

variety of bifunctional acyclic nitrogen derivatives such as diamides, dinitriles, and amidic acids.⁶ These reactions sometimes require drastic cyclization conditions which are not compatible with many functional groups.

In this paper we describe further synthetic applications of β -fragmentation reactions, promoted by alkoxy radicals generated by reaction of the corresponding carbinol amides with hypervalent iodine reagents.⁷ This methodology involves the treatment of the carbinol amides in dichloromethane with (diacetoxyiodo)benzene (DIB) and iodine under irradiation with visible light at 25 °C. These mild conditions make it possible to extend the reaction to the preparation of substituted imides not available by other methods. The synthesis of substituted succinimides from easily accessible carbinol amides of the types 1-hydroxy-2-azabicyclo[3.3.0]octan-3-one (e.g., 4, 6, and 7) and 1-hydroxy-2-azabicyclo[3.2.1]octan-3-one (e.g., 16, 17, and 21) by this method is described.

The process is expected to occur through the mechanism outlined in Scheme 1. The alkoxy radical A initially formed undergoes two types of β -fragmentation to give radicals B and C. Radical B was stabilized by trapping an iodine radical from the medium, while N-radical C

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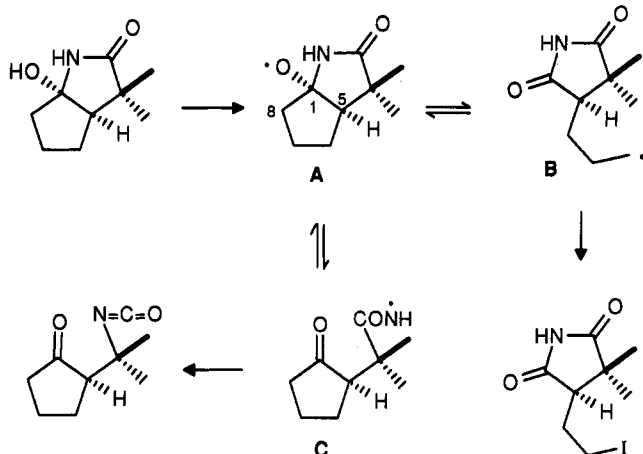
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Scheme 1



suffers an amidyl rearrangement⁸ to give the isocyanate. The other possible fragmentation through C₁-C₅ was not observed.

The intermediate carbinol amides can be prepared by a number of simple procedures from α,β -unsaturated ketones.⁹ Substrates required for this study were synthesized by Michael addition of cyanide anion to the corresponding enone followed by hydrolysis.

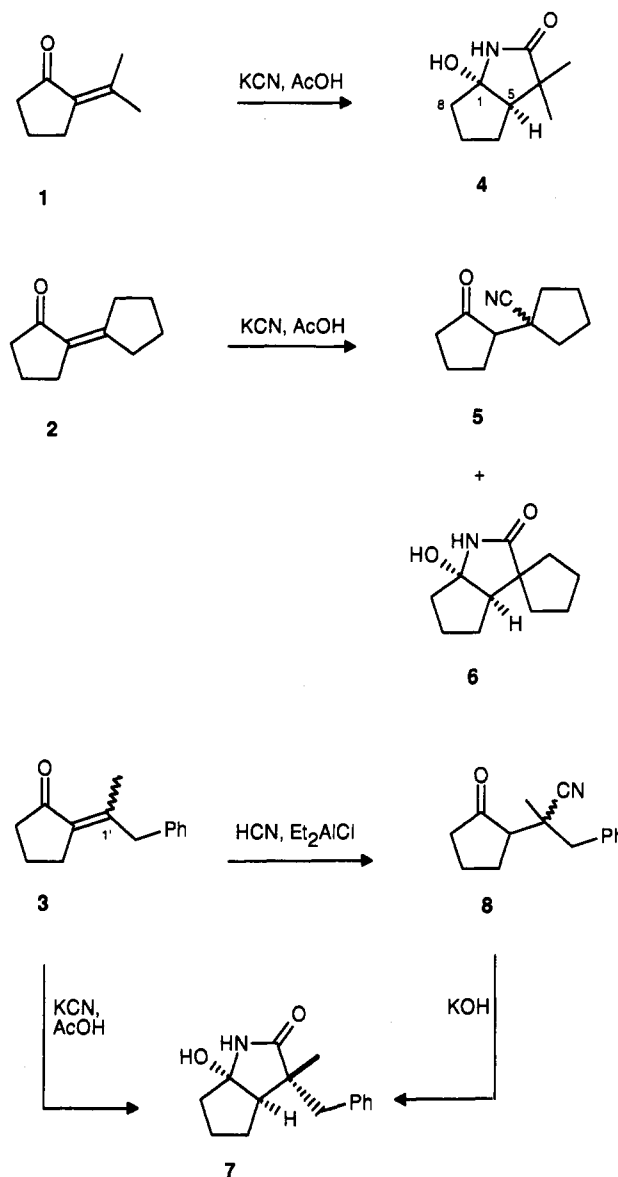
Preliminary results to ascertain the feasibility of this approach were reported earlier.¹⁰

Results and Discussion

Fragmentation of 1-Hydroxy-2-azabicyclo[3.3.0]-octan-3-ones. The α,β -unsaturated ketones were prepared essentially following previously reported procedures.¹¹ Compounds **1**¹² and **2**¹³ were obtained by aldol condensation of cyclopentanone with the corresponding ketone under basic conditions (Scheme 2). Enones **3Z** and **3E** were synthesized by a modified Mukaiyama¹⁴ aldol addition of 1-[(trimethylsilyloxy)-1-cyclopentene¹⁵ with benzyl methyl ketone in the presence of titanium tetrachloride and subsequent dehydration with *p*-toluenesulfonic acid of the resulting crude reaction mixture of tertiary alcohols in 89% overall yield. The stereochemistry of the double bond was determined by ¹H NMR spectroscopy.¹⁶ Thus, the 1'-Me in the **3Z** isomer appears at 1.74 ppm (t, *J* = 1.4 Hz), while in the **3E** isomer it is at 2.15 ppm (t, *J* = 2 Hz).

The racemic carbinol amides **4**,¹⁷ **6**, and **7** were synthesized by reaction of the corresponding enones **1**, **2**, and **3E/Z** utilizing the Lapworth procedure¹⁸ by treatment with KCN in EtOH-H₂O-AcOH in 80, 53, and 85% yield, respectively. Intermediate nitriles, e.g., **5**, were obtained

Scheme 2*



* Products **4**, **6**, and **7** are racemic although single enantiomers are shown.

in some cases when the hydrolytic process was not completed. Alternatively, the carbinol amide **7** was prepared in a two-step sequence by hydrocyanation of the enone **3E/Z** with hydrogen cyanide and diethylaluminum chloride following the procedure of Nagata *et al.*^{19,20} and subsequent hydrolysis of the nitrile intermediate **8** with 4% KOH in EtOH-H₂O solution. This latter sequence produced a poorer overall yield. It is worth noting that in all cases only one isomer of the carbinol amides **4**, **6**, and **7** was obtained. In the case of the carbinol amide **7** this can be explained considering an equilibrium in the Michael addition with hydrolysis of the cyanide **8** placing the benzyl group in the more stable *exo* position.

The β -fragmentation of carbinol amides **4**, **6**, and **7** proceeded smoothly by irradiation with visible light (two 100-W tungsten filament lamps) under mild conditions (25 °C, 45-60 min) in the presence of DIB (1.5 equiv) and

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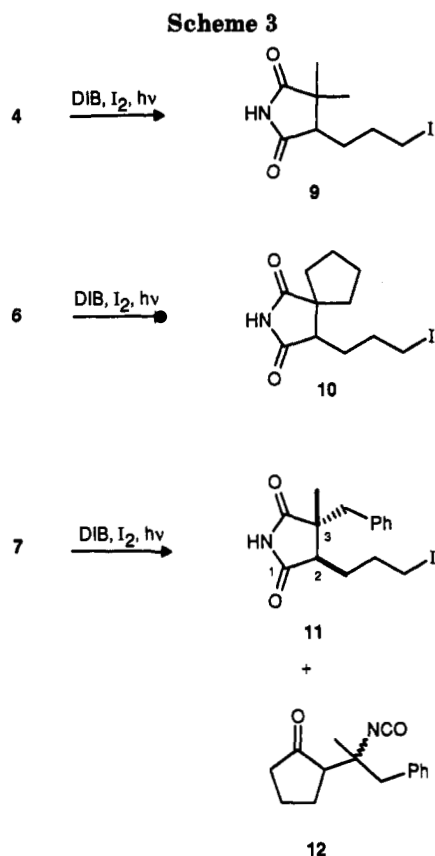
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iodine (1 equiv) in dichloromethane to give the imides 9, 10, and 11, respectively, in good yield (Scheme 3). Small amounts of isocyanates, e.g., 12, could also be isolated by chromatography of the crude reaction mixture.

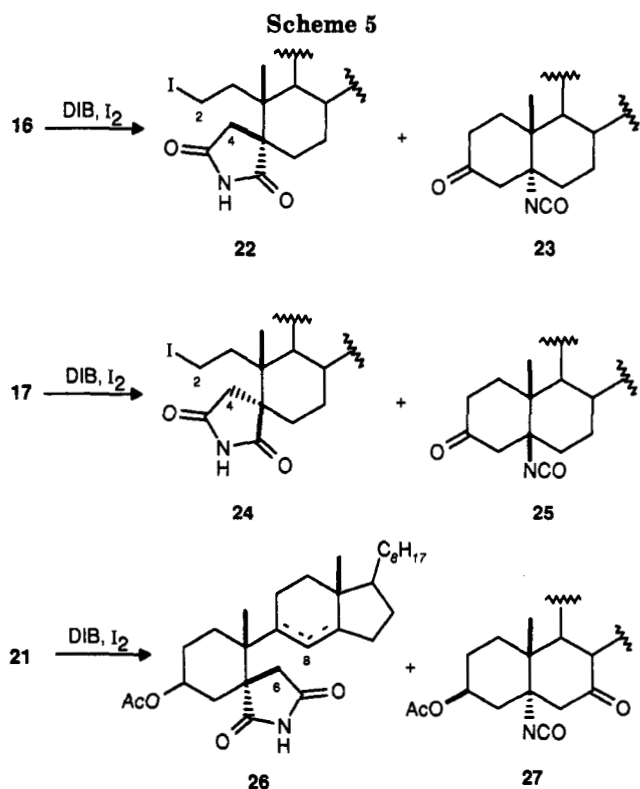
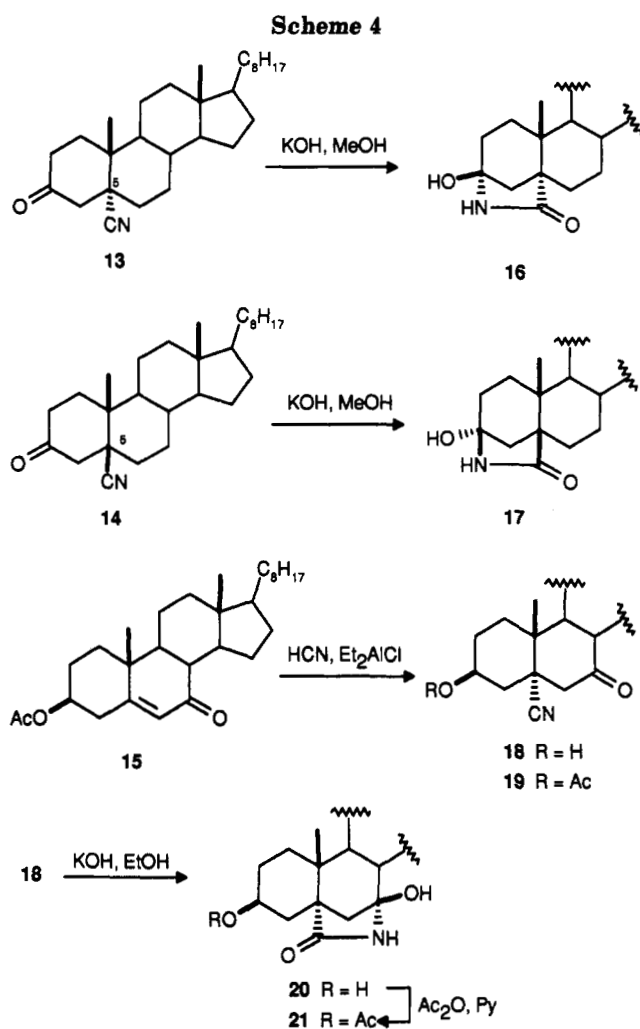
Although, as shown in Scheme 1, stabilization of the alkoxy radical A can proceed by fragmentation of C₁–C₈ or C₁–C₅ bonds to give five- or eight-membered cyclic imides via stabilization of primary or secondary C-radicals, complete regioselectivity was observed in these cases. No medium-sized imides were detected; only the more stable five-membered imides were obtained. The isocyanate 12 was presumably formed by amidyl rearrangement.⁸

Fragmentation of 1-Hydroxy-2-azabicyclo[3.2.1]-octan-3-ones. Three different steroidal carbinol amides 16, 17, and 21 were prepared by hydrocyanation of the corresponding α,β -unsaturated ketones using the method of Nagata *et al.*^{19,20} (Scheme 4). In the case of cholest-4-en-3-one a mixture of isomeric carbonitriles 13 and 14 was obtained. These were separated by chromatography and independently hydrolyzed to afford carbinol amides 16 and 17. The hydrocyanation of 3 β -acetoxycholest-5-en-7-one (15) occurred only from the α -side of the molecule to give a mixture of compounds 18 and 19. The 5 α -carbonitrile 18 was hydrolyzed and acetylated to the carbinol amide 21 in 89% overall yield.

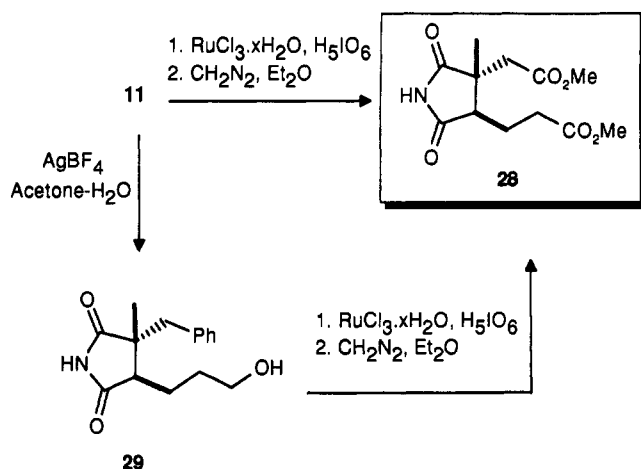
Fragmentation of these carbinol amides promoted by DIB and iodine gave the spiroimides 22, 24, and 26 in 62, 63, and 38% yield, respectively (Scheme 5). The double bond position of 26 was not determined.

Again, complete regioselectivity was observed; however, significant amounts of isocyanates 23, 25, and 27 also formed.

Synthesis of the Ring-B Imide of Vitamin B₁₂. With the succinimide 11 in hand, we then addressed the synthesis of ring-B imide 28⁴ in racemic form (Scheme 6).



This compound has attracted considerable synthetic interest because it is the essential building block in the

Scheme 6^a

^a Products **28** and **29** are racemates although only one enantiomer is shown.

syntheses of isobacteriochlorins. Compound **28**, which was first obtained in small amounts by degradation of vitamin B₁₂, was recently synthesized, first as a racemate and then in optically active form by multistep sequences.^{4,21} We used a simple and straightforward approach to prepare the 2-[2-(methoxycarbonyl)ethyl] and 3-[(methoxycarbonyl)methyl] substituents of the ring-B imide. The conversion of phenyl groups into carboxylic acids has been well established.²² Recently, we showed²³ that ruthenium tetroxide, a widely recognized oxidizing agent,²⁴ efficiently oxidizes primary alkyl iodides to carboxylic acids. Thus, when the succinimide **11** was treated with ruthenium tetroxide and the crude reaction treated with ethereal diazomethane and chromatographed, the imide **28** was obtained in 54% yield. Compound **28** can also be obtained from **11** in a 42% overall yield by a two-step sequence by treatment with silver tetrafluoroborate²⁵ and subsequent oxidation of the resulting alcohol **29** with RuO₄. The mp, IR, and ¹H and ¹³C NMR spectra of compound **28** were identical with those previously described.²¹ The overall yield of **28** for the four operations from 1-[(trimethylsilyloxy)-1-cyclopentene] → **3** → **7** → **11** → **28** was 29%. This demonstrates that this reaction sequence constitutes a very feasible synthetic route to the racemic imide **28**.

In summary, the DIB/I₂-assisted β-fragmentation of 4,4-substituted carbinol amides is a mild and simple method for the syntheses of 2-(3-iodopropanyl)-substituted succinimides. The iodine atom can be conveniently transformed into different useful functional groups, e.g., carboxylic acids, alcohols, etc. This method permitted a short and efficient synthesis of the ring-B imide **28** with good overall yield from simple commercially available starting materials. Syntheses of other biologically interesting *trans*-2,3-substituted succinimides by this method are in progress.

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Experimental Section

General. Melting points were determined with a Kofler hot-stage apparatus and are uncorrected. Optical rotation measurements were recorded at room temperature in CHCl₃ on Perkin-Elmer 141 or 142 polarimeters. IR spectra were recorded on a Perkin-Elmer 1605/FTIR spectrometer in CHCl₃ solutions. ¹H NMR (200 MHz) and ¹³C NMR (50.3 MHz) spectra (δ) were recorded in CDCl₃ on a Bruker WP 200 SY spectrometer with Me₄Si as internal standard. Low-resolution mass spectra were determined with Hewlett-Packard 5930 A or VG Micromass ZAB-2F spectrometers and high-resolution mass spectra on a VG Micromass ZAB-2F spectrometer. Merck silica gel 60 PF₂₅₄ and 60 (0.063–0.2 mm) were used for preparative thin-layer chromatography (TLC) and column chromatography, respectively. Circular layers of 1 mm of Merck silica gel 60 PF₂₅₄ were used on a Harrison Chromatotron for centrifugally assisted chromatography. Commercial reagents and solvents were analytical grade or were purified by standard procedures prior to use. The spray reagent for TLC was vanillin (1 g) in H₂SO₄-EtOH (4:1, 200 mL). (Diacetoxyiodo)benzene (DIB) 98% was purchased from Aldrich. 2-Isopropylidene-cyclopentanone (**1**)¹² and 2-cyclopentylidene-cyclopentanone (**2**)¹³ were prepared following previously reported procedures in 40 and 60% yield, respectively.

2-(1-Methyl-2-phenylethylidene)cyclopentanone (3). A solution of 1-[(trimethylsilyloxy)-1-cyclopentene] (12 g, 0.77 mol) in CH₂Cl₂ (300 mL) was added dropwise to a stirred solution of benzyl methyl ketone (10.8 g, 0.81 mol) and TiCl₄ (15.36 g, 0.81 mol) in dry CH₂Cl₂ (150 mL), over 3–4 Å molecular sieves, under an argon atmosphere and at -78 °C. The resulting solution was allowed to reach room temperature and the stirring continued for 7 h, and then the solution was poured into H₂O (500 mL) and extracted with CH₂Cl₂ (3 × 50 mL). The combined organic layers were dried, filtered, and concentrated. The crude mixture in C₆H₆ (100 mL) was treated with *p*-toluenesulfonic acid (190 mg, 1 mmol) for 16 h at rt. To the mixture was added H₂O (200 mL), and it was extracted with CH₂Cl₂ (3 × 50 mL). The organic phase was washed with 5% NaOH (2 × 50 mL), 5% HCl (50 mL), and H₂O (50 mL), dried, and evaporated. The crude products were purified by column chromatography (hexane-ethyl acetate, 98:2) to yield **3** (13.69 g, 89%) as a mixture of *Z*- and *E*-olefins (2:1, ¹H NMR analysis). Chromatotron chromatography of a sample (233 mg) (C₆H₆-EtOAc, 99.7:0.3) gave **3Z** enone (151 mg) and **3E** enone (71 mg). Compound **3Z**: amorphous; IR 1699, 1625, 1601, 1495, 1453, 1410, 1373 cm⁻¹; ¹H NMR 1.74 (3H, t, *J* = 1.4 Hz), 1.89 (2H, quintuplet, *J* = 7.5 Hz), 2.39 (2H, t, *J* = 8 Hz), 2.62 (2H, m), 4.13 (2H, br s), 7.1–7.4 (5H, m); ¹³C NMR 19.13 (t), 21.66 (q), 29.62 (t), 38.20 (t), 40.72 (t), 125.99 (d), 128.26 (2 × d), 128.99 (2 × d), 131.57 (s), 139.58 (s), 148.56 (d), 207.38 (s); MS *m/z* (rel intensity) 200 (M⁺, 100), 185 (16), 183 (30), 129 (63), 91 (72); HRMS calcd for C₁₄H₁₆O 200.1201, found 200.1184. Compound **3E**: amorphous; IR 1701, 1624, 1601, 1495, 1453, 1410, 1372 cm⁻¹; ¹H NMR 1.91 (2H, quintuplet, *J* = 7.4 Hz), 2.15 (3H, t, *J* = 2 Hz), 2.37 (2H, t, *J* = 7.9 Hz), 2.75 (2H, m), 3.47 (2H, br s), 7.1–7.4 (5H, m); ¹³C NMR 18.48 (q), 19.56 (t), 29.62 (t), 40.59 (t), 43.81 (t), 126.44 (d), 128.58 (2 × d), 128.63 (2 × d), 132.00 (s), 137.97 (s), 148.33 (s), 208.06 (s); MS *m/z* (rel intensity) 200 (M⁺, 100), 185 (14), 183 (25), 129 (61), 91 (78); HRMS calcd for C₁₄H₁₆O 200.1201, found 200.1190.

(±)-(1*S*,5*R*)-4,4-Dimethyl-1-hydroxy-2-azabicyclo[3.3.0]octan-3-one (4). To a solution of enone **1** (30 g, 0.242 mol) in ethanol (50 mL) was added a solution of KCN (16.15 g, 0.248 mol) in EtOH (700 mL) and water (38 mL). To the resulting mixture was added dropwise acetic acid (4.5 mL, 75 mmol) in EtOH (250 mL) and the mixture stirred at room temperature for 24 h and then poured into water (2 L) and extracted with CH₂Cl₂ (3 × 250 mL). The organic phase was washed with aqueous NaHCO₃ (2 × 50 mL) and water (50 mL), dried, and evaporated. The residue was purified by crystallization from EtOAc to give **4** (32.7 g, 80%): mp 118–120 °C (lit.¹⁷ mp 117.5–118.5 °C); IR 3585, 3410, 1685 cm⁻¹; ¹H NMR 1.04 (3H, s), 1.29 (3H, s), 1.50–1.95 (6H, m), 2.23 (1H, m), 5.04 (1H, br s), 7.45 (1H, br s); ¹³C NMR 20.70 (q), 24.97 (t), 28.38 (t), 28.86 (q), 40.48 (t), 42.58 (s), 56.65 (d), 95.29 (s), 183.23 (s); MS *m/z* (rel intensity) 169 (M⁺, 12), 154 (16), 151 (88), 136 (100), 127 (18), 122 (53), 111 (47), 108 (83).

23.89 (t), 25.08 (t), 26.40 (t), 28.09 (d), 28.54 (t), 31.17 (t), 35.75 (d), 36.24 (t), 38.83 (t), 39.59 (t), 39.82 (t), 39.88 (s), 42.92 (s), 48.95 (d), 49.07 (d), 50.41 (d), 52.67 (t), 55.02 (d), 69.67 (d), 70.05 (s), 123.29 (s), 170.30 (s), 207.96 (s); MS *m/z* (rel intensity) 485 (M^+ , 90), 470 (6), 467 (33), 443 (18), 424 (15), 382 (100), 364 (21), 277 (36), 228 (30); HRMS calcd for $C_{30}H_{47}NO_4$ 485.3505, found 485.3480. Compound **26**: mp 157–158 °C (from MeOH), $[\alpha]_D^{25} = +27^\circ$ ($CHCl_3$, $c = 0.26$); IR 3395, 1765, 1720 cm^{-1} ; 1H NMR 0.49 (3H, s), 0.86 (6H, d, $J = 6.5$ Hz), 0.87 (3H, d, $J = 6.4$ Hz), 1.17 (3H, s), 2.19 (1H, d, $J = 18.4$ Hz), 2.96 (1H, d, $J = 18.4$ Hz), 5.37 (1H, m), 2.01 (3H, s), 5.62 (1H, m), 7.59 (1H, m); ^{13}C NMR 11.02 (q), 18.20 (q), 20.19 (q), 21.17 (q), 22.49 (q), 22.71 (q), 23.63 (t), 26.55 (t), 27.96 (d), 28.21 (t), 29.08 (t), 29.63 (t), 31.21 (t), 35.70 (d), 36.15 (t), 37.62 (t), 39.51 (t), 39.93 (s), 41.47 (t), 42.26 (s), 42.57 (t), 46.60 (d), 52.54 (s), 56.23 (d), 69.16 (d), 125.73 (d), 138.70 (s), 170.04 (s), 174.99 (s), 180.52 (s); MS *m/z* (rel intensity) 485 (M^+ , 48), 470 (4), 425 (56), 410 (16), 372 (25), 312 (100), 247 (80); HRMS calcd for $C_{30}H_{47}NO_4$ 485.3505, found 485.3501.

(±)-(2*R**,3*R**)-2-[2-(Methoxycarbonyl)ethyl]-3-[(methoxycarbonyl)methyl]-3-methylsuccinimide (Ring-B Imide) (**28**). **Method A.** Periodic acid (1.02 g, 4.4 mmol, 22.4 equiv) was added to a stirred solution of **11** (74.2 mg, 0.2 mmol) in carbon tetrachloride (0.5 mL), acetonitrile (0.5 mL), and water (0.75 mL), under positive pressure of dry Ar. After 15 min, ruthenium trichloride hydrate (1 mg, 0.02 equiv) was added and the stirring continued for 6 h at room temperature. The reaction mixture was cooled to 0 °C with an ice bath, and ether (2 mL) was added. Vigorous stirring was continued for 10 min, and the product was then extracted with ether (3 × 10 mL). The combined organic layers were washed with brine (3 × 10 mL) and water (10 mL), dried, filtered, and concentrated. The crude product dissolved in Et_2O (3 mL) was treated with excess ethereal diazomethane and purified by chromatotron chromatography (hexane–EtOAc, 7:3) to give the dimethyl ester imide **28** (29.3 mg, 54%): mp 115–115.5 °C (EtOAc–hexane) (lit.²⁰ mp 111.5–113.5 °C); IR 3401, 1782, 1729, 1719 cm^{-1} ; 1H NMR 1.23 (3H, s), 1.7–2.1 (2H, m), 2.54–2.84 (2H, m), 2.63 (1H, d, $J = 17.6$ Hz), 2.90

(1H, d, $J = 17.6$ Hz), 2.96 (1H, dd, $J = 9.5, 4.2$ Hz), 3.68 (3H, s), 3.69 (3H, s), 9.02 (1H, br s); 1H NMR (C_6D_6) 0.67 (3H, s), 1.34–1.79 (2H, m), 2.20 (1H, d, $J = 17.5$ Hz), 2.59 (1H, d, $J = 17.5$ Hz), 2.44 (2H, m), 2.96 (1H, dd, $J = 9.1, 5.1$ Hz), 3.21 (3H, s), 3.28 (3H, s), 7.97 (1H, br s); ^{13}C NMR 20.45 (t), 20.63 (q), 32.03 (t), 39.31 (t), 45.99 (s), 46.96 (d), 51.59 (q), 51.87 (q), 171.11 (s), 173.27 (s), 178.76 (s), 181.55 (s); MS *m/z* (rel intensity) 271 (M^+ , 5), 240 (75), 211 (33), 198 (41), 185 (40), 180 (34), 166 (100); HRMS calcd for $C_{12}H_{17}NO_6$ 271.1056, found 271.1078.

Method B. Silver tetrafluoroborate (204 mg, 1.05 mmol) was added to a solution of **11** (259 mg, 0.7 mmol) in acetone (9 mL) and water (1 mL), and the mixture was stirred in the dark for 30 min, at room temperature, and then poured into water (30 mL) and extracted with CH_2Cl_2 (3 × 15 mL). The combined CH_2Cl_2 layers were dried, filtered, and evaporated. Chromatography of the residue (hexane–EtOAc, 7:3) gave the hydroxyimide **29** (151 mg, 83%): amorphous; IR 3614, 3401, 1776, 1714, 1600, 1495, 1453, 1380 cm^{-1} ; 1H NMR 1.31 (3H, s), 1.60–1.90 (4H, m), 2.29 (1H, m), 2.64 (1H, d, $J = 13.8$ Hz), 3.27 (1H, d, $J = 13.8$ Hz), 2.64 (1H, t, $J = 6.7$ Hz), 3.63 (2H, m), 7.1–7.3 (5H, m), 8.88 (1H, m); ^{13}C NMR 21.05 (q), 21.88 (t), 30.95 (t), 42.19 (t), 46.24 (d), 50.15 (s), 61.98 (t), 127.06 (d), 128.55 (2 × d), 130.13 (2 × d), 136.25 (s), 179.92 (s), 182.67 (s); MS *m/z* (rel intensity) 261 (M^+ , 3), 243 (3), 231 (2), 202 (7), 170 (9), 152 (13), 91 (100); HRMS calcd for $C_{15}H_{19}NO_3$ 261.1364, found 261.1362. A solution of **29** (124 mg, 0.48 mmol) in CCl_4 (1 mL), CH_3CN (1 mL), and H_2O (1.5 mL) when treated with H_5IO_6 (1.86 g, 8.1 mmol, 14 equiv) and $RuCl_3 \cdot xH_2O$ (2.4 mg, 0.02 equiv) as described previously for **11** afforded **28** (65.7 mg, 51%).

Supplementary Material Available: 1H NMR and ^{13}C NMR spectra of compounds **3Z**, **3E**, **5–16**, **19–26**, **28**, and **29** and 1H NMR spectrum of compound **27** (49 pages). This material is contained in libraries on microfiche, immediately follows this article in the microfilm version of the journal, and can be ordered from the ACS; see any current masthead page for ordering information.